

Conversion of Various Lab Analysis to Phosphorus ppm Mehlich I (MI)

Waypoint Analytical (Formerly A&L)

Lab	Analysis	Conversion Needed	P ppm Mehlich I
Waypoint	49 P-ppm	$(49 \text{ ppm} \times 0.458) - 3.26$	= 19

- We want our analysis is PPM Mehlich I
- Waypoint Analytical uses Mehlich III Extraction
- $(\text{Mehlich III} \leq 205 \text{ ppm}) \times 0.485 - 3.26 = \text{Mehlich 1 ppm}$

$$A. 49 \times .458 = 22.442$$

$$B. 22.442 - 3.26 = 19$$

Lab	Analysis	Conversion Needed	P ppm Mehlich I
Brookside	256 P ₂ O ₅ Lbs./ac	256 P ₂ O ₅ lbs./ac X 0.22 56.32 ppm X .458 – 3.26	= 23

- We want our analysis in PPM Mehlich I.
- Brookside uses Mehlich III.
- P₂O₅ lbs./ac X 0.22 = P ppm

A. 256 P₂O₅ lbs./ac X 0.22 = 56.32 ppm P

*after conversion analysis is below 205

B. 56.32 ppm X .458 – 3.26

Lab	Analysis	Conversion Needed	P ppm Mehlich I
Spectrum	214 P - ppm	$(214 \text{ ppm} \times 0.945) - 103.5$	$= 99$

- We want our analysis is PPM Mehlich I.
- Spectrum uses Mehlich III.
- $(\text{Mehlich III} \geq 206 \text{ ppm}) \times 0.945 - 103.5 = \text{Mehlich 1 ppm}$

$$\text{A. } 214 \text{ ppm} \times 0.945 - 103.5 = 99$$

Lab	Analysis	Conversion Needed	P ppm Mehlich I
VA Tech	6 P – lbs./ac	6 lbs./ac X 0.5	= 3

- We want our analysis is PPM Mehlich I
- VA Tech uses Mehlich I

$$P \text{ lbs./ac} \times 0.5 = P \text{ ppm}$$

$$A. 6 \text{ lbs./ac} \times 0.5 = 3$$